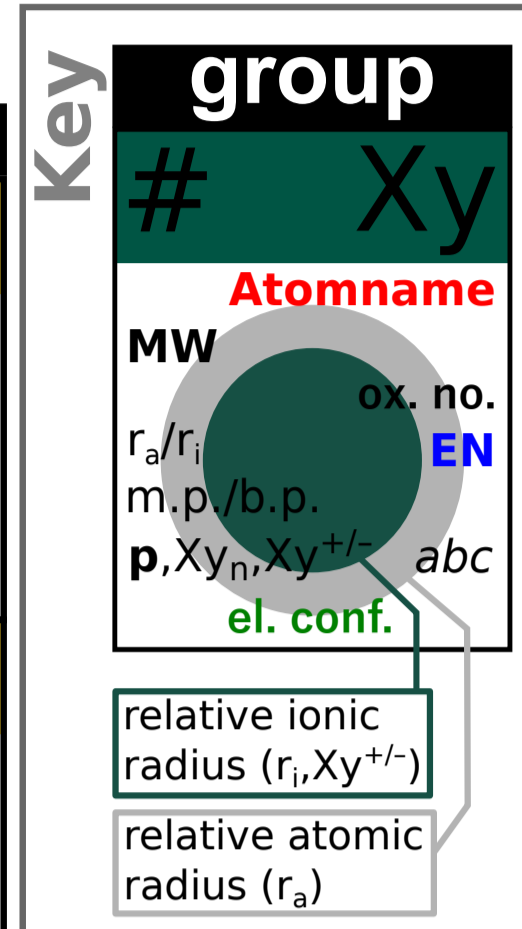


Periodic Table of the Elements

1-IA
1 H Hydrogen 1.00794 -1,1 52.9/154 13.99/20.271 g, H ₂ , H hex 1s ²
2-IIA
3 Li Lithium 6.941 1 167/90 453.65/1603 s, Li ⁺ , Li ⁺ bcc 1s ² 2s ¹
4 Be Beryllium 9.0121831(5) 2 112/59 1560/2742 s, Be _n , Be ²⁺ hcp 1s ² 2s ²
11 Na Sodium 22.98976928(2) 1 190/116 370.94/1156.09 s, Na _n , Na ⁺ bcc [Ne]3s ¹
12 Mg Magnesium 24.305 2 145/86 923/1363 s, Mg _n , Mg ²⁺ hcp [Ne]3s ²
19 K Potassium 39.0983(1) 1 243/152 336.7/1032 s, K _n , K ⁺ bcc [Ar]4s ¹
20 Ca Calcium 40.078(4) 2 194/114 1115/1757 s, Ca _n , Ca ²⁺ fcc [Ar]4s ²
37 Rb Rubidium 85.4678(3) 1 265/166 312.45/961 s, Rb _n , Rb ⁺ bcc [Kr]5s ¹
38 Sr Strontium 87.62(1) 2 219/132 1050/1650 s, Sr _n , Sr ²⁺ fcc [Kr]5s ²
55 Cs Cesium 132.90545196 1 298/181 301.7/944 s, Cs _n , Cs ⁺ bcc [Xe]6s ¹
56 Ba Barium 137.327(7) 2 253/149 1000/2118 s, Ba _n , Ba ²⁺ bcc [Xe]6s ²
87 Fr Francium (223) 1 n.a./n.a. n.a./n.a. n.a./n.a. n.a. bcc [Rn]7s ¹

2-IIA
9 Be Beryllium 9.0121831(5) 2 112/59 1560/2742 s, Be _n , Be ²⁺ hcp 1s ² 2s ²
12 Mg Magnesium 24.305 2 145/86 923/1363 s, Mg _n , Mg ²⁺ hcp [Ne]3s ²
20 Ca Calcium 40.078(4) 2 194/114 1115/1757 s, Ca _n , Ca ²⁺ fcc [Ar]4s ²
38 Sr Strontium 87.62(1) 2 219/132 1050/1650 s, Sr _n , Sr ²⁺ fcc [Kr]5s ²
56 Ba Barium 137.327(7) 2 253/149 1000/2118 s, Ba _n , Ba ²⁺ bcc [Xe]6s ²
88 Ra Radium (226) 2 n.a./162 973/2010 s, Ra _n , Ra ²⁺ bcc [Rn]7s ²



Note: values written in gray are predictions
#.....Atomic number
Xy.....Symbol
Atomname.....If written in black, the atom is usually produced synthetically.
MW.....Molecular weight (g/mol)
ox. no.Most common oxidation states
EN.....Electro negativity (pauling scale)
r_a.....atomic radius (pm)
r_i.....ionic radius (pm)
m.p.melting point (K)*
b.p.boiling point (K)*
pphases*: solid (s), liquid (l), gas (g)
Xy.....Elementar form
Xy^{+/-}.....ion corresponding to r_i
el. conf.electron configuration
abc.....crystal structure
*Values at STP (273.15 K, 1 bar)

Equations:
Concentration: c = n/V [mol/L]
Amount of substance: n [mol]
Volume: V [L]
Particle number: N = n · N_A
Pressure: p [Pa]
Ideal gas equation: pV = nRT = Nk_BT
bcc: body centered cubic
cub: cubic
dhcp: double hexagonal close-packed
fcc: face-centered cubic
fcd: face-centered diamond-cubic
hcp: hexagonal closed-packed
hex: hexagonal
mon: monoclinic
ort: orthorhombic
rho: rhombohedral
she: simple hexagonal

Conversion factors:
1 μm = 10⁻⁶ m; 1 nm = 10⁻⁹ m; 1 Å (Angs.) = 10⁻¹⁰ m; 1 pm = 10⁻¹² m; 1 fm = 10⁻¹⁵ m
1 bar = 10⁵ N/m² = 10⁵ Pa; 1 atm = 101325 Pa = 1.01325 bar
Torr = 1/760 atm = 1.333 mbar = 1 mmHg
1 L = 10⁻³ m³ = 1 dm³ = 10³ cm³ = 10⁶ mm³
Constants:
Avogadro number N_A = 6.022 141 79(30) · 10²³ mol⁻¹
Mass of proton m_p = 1.672 621 777(74) · 10⁻²⁷ kg
Mass of electron m_e = 9.109 382 91(40) · 10⁻³¹ kg
Mass of neutron m_n = 1.674 927 351(74) · 10⁻²⁷ kg
Universal gas constant R = 8.314 472(15) J/(mol·K)
Boltzmann-constant k_B = 1.380 650 4(24) · 10⁻²³ J/K
Speed of light c = 2.997 924 58 · 10⁸ m/s
Elementary charge e = 1.602 176 487(40) · 10⁻¹⁹ C
Planck constant h = 6.626 068 96(33) · 10⁻³⁴ J·s
Unified atomic mass unit 1 u = 1.660 538 921(73) · 10⁻²⁷ kg
The unified atomic mass is equal to 1/12 of the mass of a single isolated C-atom.

REFERENCES:
[MW] Comission on Isotopic Abundancies and Atomic Weights, http://www.ciaaw.org/
[r_a] E. Clementi, D.L. Raimondi, W.P. Reinhardt, *J. Chem. Phys.*, **1967**, *47*, 1300-1307.
[r_i] R. D. Shannon, *Acta Cryst.*, **1976**, *A32*, 751-767 and https://en.wikipedia.org/wiki/ionic_radius.
[m.s., b.p., phases, cryst. struct., ox. no.] https://www.wikipedia.org
[EN] A. L. Allred, *J. Inorg. Nucl. Chem.*, **1961**, *17*, 215-221.
[Constants] http://physics.nist.gov/cuu/Constants/index.html

18-VIIIB
2 He Helium 4.002602(2) 0 31/- 0.95/4.222 g, He 1s ²
10 Ne Neon 20.1797(6) 0 38/- 24.56/27.104 g, Ne 1s ² 2s ² 2p ⁶
18 Ar Argon 39.948(1) 0 71/- 83.81/87.302 g, Ar [Ne]3s ² 3p ⁶
36 Kr Krypton 83.798(2) 0 115/- 119.93 g, Kr [Ar]3d ¹⁰ 4s ² 4p ⁶
54 Xe Xenon 131.293(6) 0 108 161.40/165.051 g, Xe [Kr]4d ¹⁰ 5s ² 5p ⁶
86 Rn Radon (222) 0 120 202/211.5 g, Rn [Xe]4f ¹⁴ 5d ¹⁰ 6s ² 6p ⁶

57 La Lanthanum 138.90547(7) 3 n.a./117.2 1193/3737 s, La _n , La ³⁺ dhcp [Xe]5d ¹ 6s ²	58 Ce Cerium 140.116(1) 3,4 n.a./101 1068/3716 s, Ce _n , Ce ⁴⁺ dhcp [Xe]4f ¹ 5d ¹ 6s ²	59 Pr Praseodymium 140.90766(2) 3 n.a./113 1208/3403 s, Pr _n , Pr ³⁺ dhcp [Xe]4f ³ 6s ²	60 Nd Neodymium 144.242(3) 3 n.a./112.3 1297/3347 s, Nd _n , Nd ³⁺ dhcp [Xe]4f ⁴ 6s ²	61 Pm Promethium (145) 3 n.a./111 1315/3273 s, Pm _n , Pm ³⁺ dhcp [Xe]4f ⁵ 6s ²	62 Sm Samarium 150.36(2) 3 n.a./109.8 1099/1802 s, Sm _n , Sm ³⁺ rho [Xe]4f ⁶ 6s ²	63 Eu Europium 151.964(1) 2,3 n.a./108.7 1585/3273 s, Eu _n , Eu ³⁺ bcc [Xe]4f ⁷ 6s ²	64 Gd Gadolinium 157.25(3) 3 n.a./107.8 1629/3396 s, Gd _n , Gd ³⁺ hcp [Xe]4f ⁷ 5d ¹ 6s ²	65 Tb Terbium 158.92535(2) 3 n.a./106.3 1629/3396 s, Tb _n , Tb ³⁺ hcp [Xe]4f ⁹ 6s ²	66 Dy Dysprosium 162.500(1) 3 n.a./105.2 1680/2840 s, Dy _n , Dy ³⁺ hcp [Xe]4f ¹⁰ 6s ²	67 Ho Holmium 164.93033(2) 3 n.a./104.1 1734/2873 s, Ho _n , Ho ³⁺ hcp [Xe]4f ¹¹ 6s ²	68 Er Erbium 167.259(3) 3 n.a./103 1818/2223 s, Er _n , Er ³⁺ hcp [Xe]4f ¹² 6s ²	69 Tm Thulium 168.93422(2) 3 n.a./102 1907/1469 s, Tm _n , Tm ³⁺ hcp [Xe]4f ¹³ 6s ²	70 Yb Ytterbium 173.045(10) 3 n.a./100.8 1097/1469 s, Yb _n , Yb ³⁺ fcc [Xe]4f ¹⁴ 6s ²	71 Lu Lutetium 174.9668(1) 3 n.a./100.1 1925/3675 s, Lu _n , Lu ³⁺ hcp [Xe]4f ¹⁴ 6s ²
89 Ac Actinium (227) 3 n.a./126 n.a./n.a. s, Ac _n , Ac ³⁺ fcc [Rn]6d ¹ 7s ²	90 Th Thorium 232.0377(4) 4 n.a./108 2023/5061 s, Th _n , Th ⁴⁺ fcc [Rn]6d ² 7s ²	91 Pa Protactinium 231.03588(2) 5 n.a./104 1841/4300 s, Pa _n , Pa ⁴⁺ tet [Rn]5f ² 6d ¹ 7s ²	92 U Uranium 238.02891(3) 6 n.a./103 1405.3/4404 s, U _n , U ⁴⁺ ort [Rn]5f ³ 6d ¹ 7s ²	93 Np Neptunium (237) 5 n.a./89 912/4474 s, Np _n , Np ⁵⁺ ort [Rn]5f ⁴ 6d ¹ 7s ²	94 Pu Plutonium (244) 4 n.a./100 912.5/3505 s, Pu _n , Pu ⁴⁺ mon [Rn]5f ⁶ 7s ²	95 Am Americium (243) 3 n.a./111.5 1449/2255 s, Am _n , Am ³⁺ dhcp [Rn]5f ⁷ 7s ²	96 Cm Curium (247) 3 n.a./99 1613/3383 s, Cm _n , Cm ⁴⁺ dhcp [Rn]5f ⁶ 6d ¹ 7s ²	97 Bk Berkelium (247) 3 n.a./110 1259/2900 s, Bk _n , Bk ³⁺ dhcp [Rn]5f ⁷ 7s ²	98 Cf Californium (251) 3 n.a./109 1173/1343 s, Cf _n , Cf ³⁺ dhcp [Rn]5f ¹⁰ 7s ²	99 Es Einsteinium (252) 3 n.a./92.8 1133/1359 s, Es _n , Es ³⁺ fcc [Rn]5f ¹¹ 7s ²	100 Fm Fermium (257) 3 n.a./n.a. 1800/n.a. s, n.a., n.a. n.a. [Rn]5f ¹² 7s ²	101 Md Mendelevium (258) 3 n.a./n.a. 1100/n.a. s, n.a., n.a. n.a. [Rn]5f ¹³ 7s ²	102 No Nobelium (259) 3 n.a./n.a. 1100/n.a. s, n.a., n.a. n.a. [Rn]5f ¹⁴ 7s ²	103 Lr Lawrencium (266) 3 n.a./n.a. 1900/n.a. s, n.a., n.a. n.a. [Rn]5f ¹⁴ 7s ² 7p ¹

